## **Rubber Oligomers**



#### SVP Symposium, March 31, 2022

**Presenter:** Dr. Piet Christiaens, Scientific Director Nelson Labs Europe

#### **Content – Rubber Oligomers**

- 1. Structure
- 2. Formation
- 3. Detection & Identification
- 4. Reactivity
- 5. Toxicity

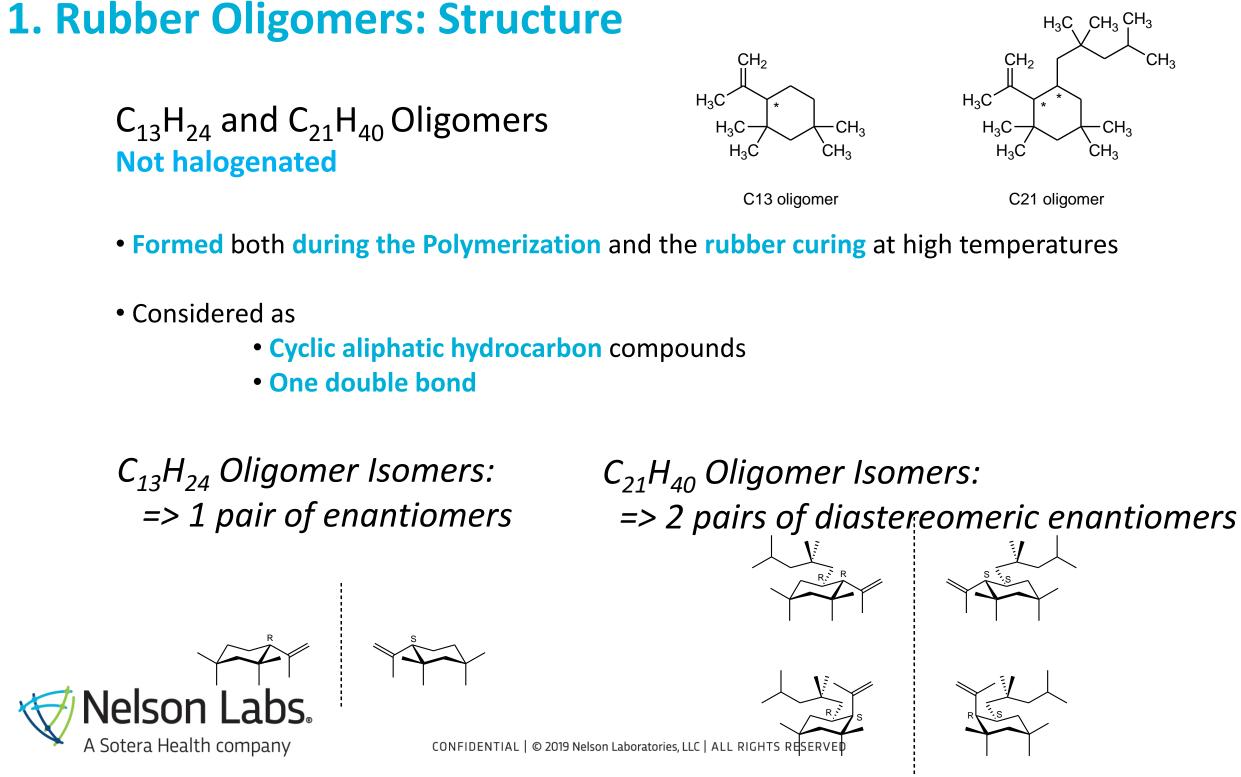




#### **1. Rubber Oligomers: Structure**







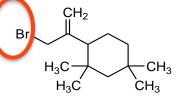
#### **1. Rubber Oligomers: Structure**

#### Halogenated Rubber Oligomers – Compounds of high concern

 $CH_2$ 

CH2

 $C_{13}H_{23}Br/C_{13}H_{23}Cl$  and  $C_{21}H_{39}Br/C_{21}H_{39}Cl$  Oligomers



- Considered as
  - HALOGENATED Cyclic Aliphatic Hydrobarbon compounds (Allyl Halide)
  - Alkylating Agents
  - One double bond
- Structure Activity Relationship (SAR) Assessment:

**CARCINOGENICITY IN HUMANS IS PLAUSIBLE** 

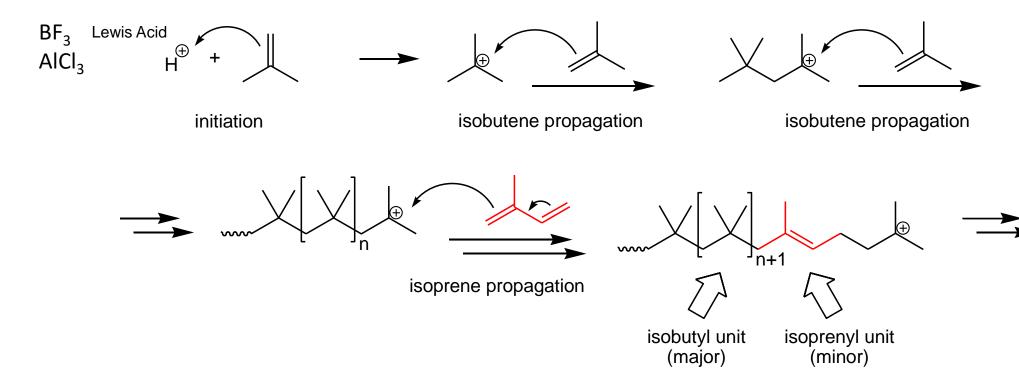
• As no experimental Literature data are known about the toxicity of these compounds







Formation (polymerization) of a Butyl Elastomer (IIR): Cationic Polymerization



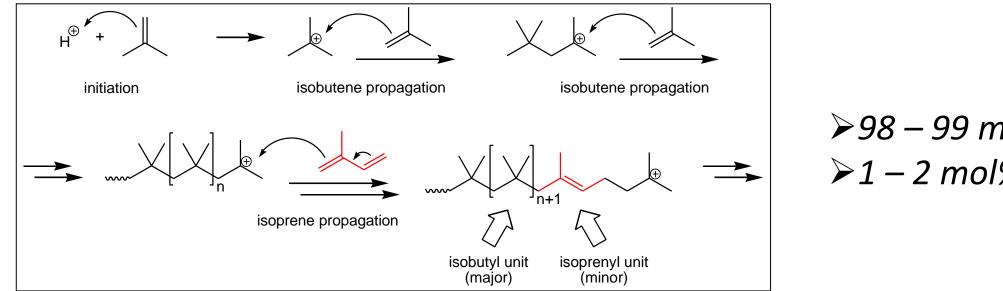
> Note: the Polymerization Starts with an isobutene Unit (present in high excess compared to isoprene!!)

o 98 – 99 mol% is isobutylene

 $\circ$  1 – 2 mol% is isoprene



Formation (polymerization) of a Butyl Elastomer (IIR): Cationic Polymerization



Means for **Butyl Elast(IIR)** that approx. **per 100 C-C bonds** in the backbone, **1** is a double (C=C) bond (if 2%) Compared with **Polyisoprene**: **Per 100 C-C bonds** in the backbone, approx. **33 will be double** (C=C) bonds

Less double bonds in IIR means:

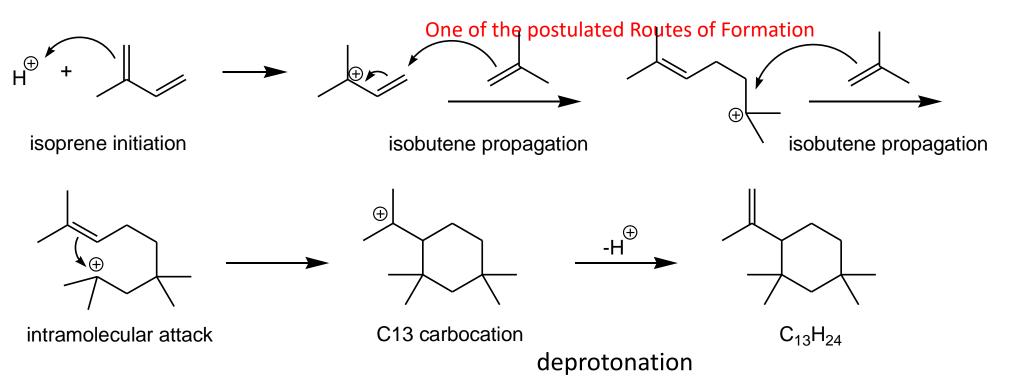
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Butyl Elastomer (IIR) is less prone to Oxidation/Ageing

Butyl Elastomer (IIR) needs a more efficient crosslinking reaction compared to Polyisoprene <u>Bromination of the backbone helps to address this (Br is a good leaving group)</u>

#### >98 – 99 mol% is isobutylene > 1 - 2 mol% is isoprene

#### Formation of the C<sub>13</sub>H<sub>24</sub> Oligomer:



#### Also: Backbite reaction during curing (not shown here)

> Note: the polymerization starts with an isoprene unit which provides the double bond which is necessary for the intramolecular ring formation

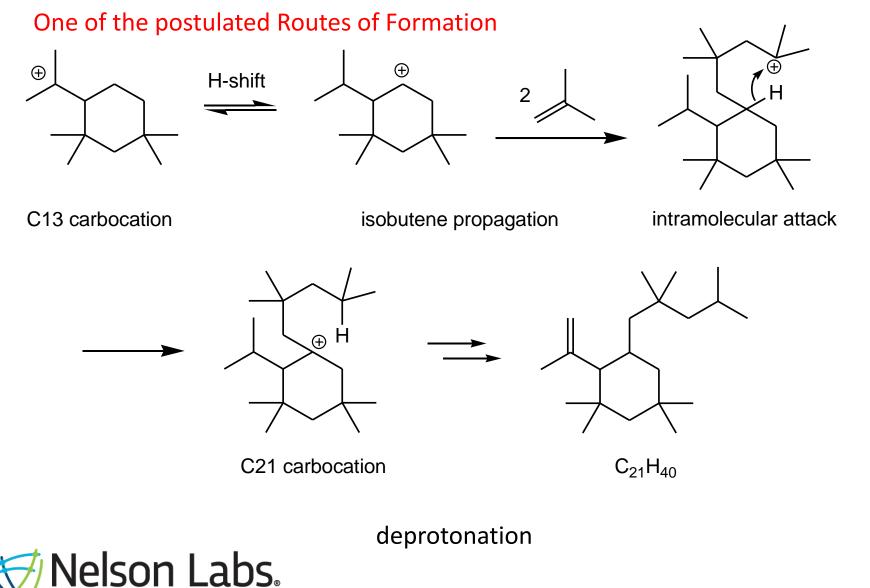




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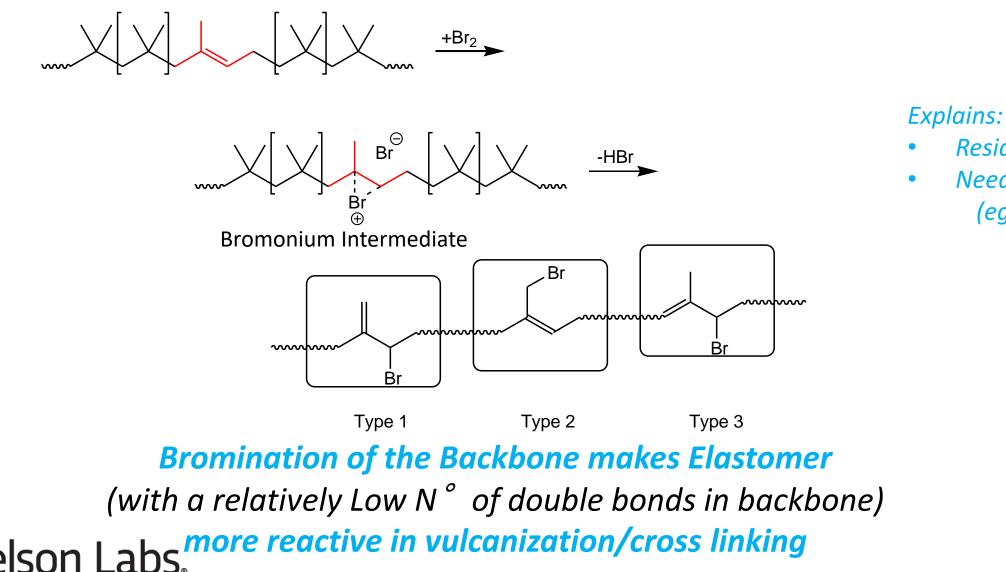
Ref. 2, 3

## Formation of the C<sub>21</sub>H<sub>40</sub> Oligomer:





**Bromobutyl Elastomer:** Bromination of a Butyl Elastomer (BIIR)

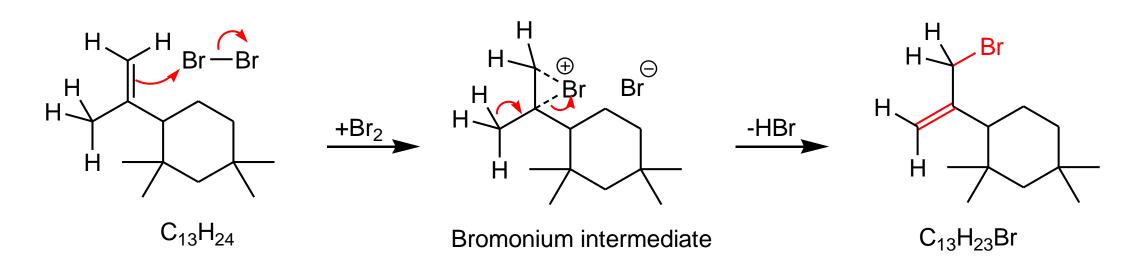




Residual Bromide in rubber *Need for Acid Scavenger* (eg Ca-Stearate)

## Formation of C<sub>13</sub>H<sub>23</sub>Br, C<sub>13</sub>H<sub>23</sub>Cl, C<sub>21</sub>H<sub>39</sub>Br & C<sub>21</sub>H<sub>39</sub>Cl:

side reaction in the production of halobutyl elastomer (halogenation of butyl elastomer)



> Note: the halogenation occurs according to classical rubber halogenation chemistry by the substitution of one of the allylic hydrogen atoms





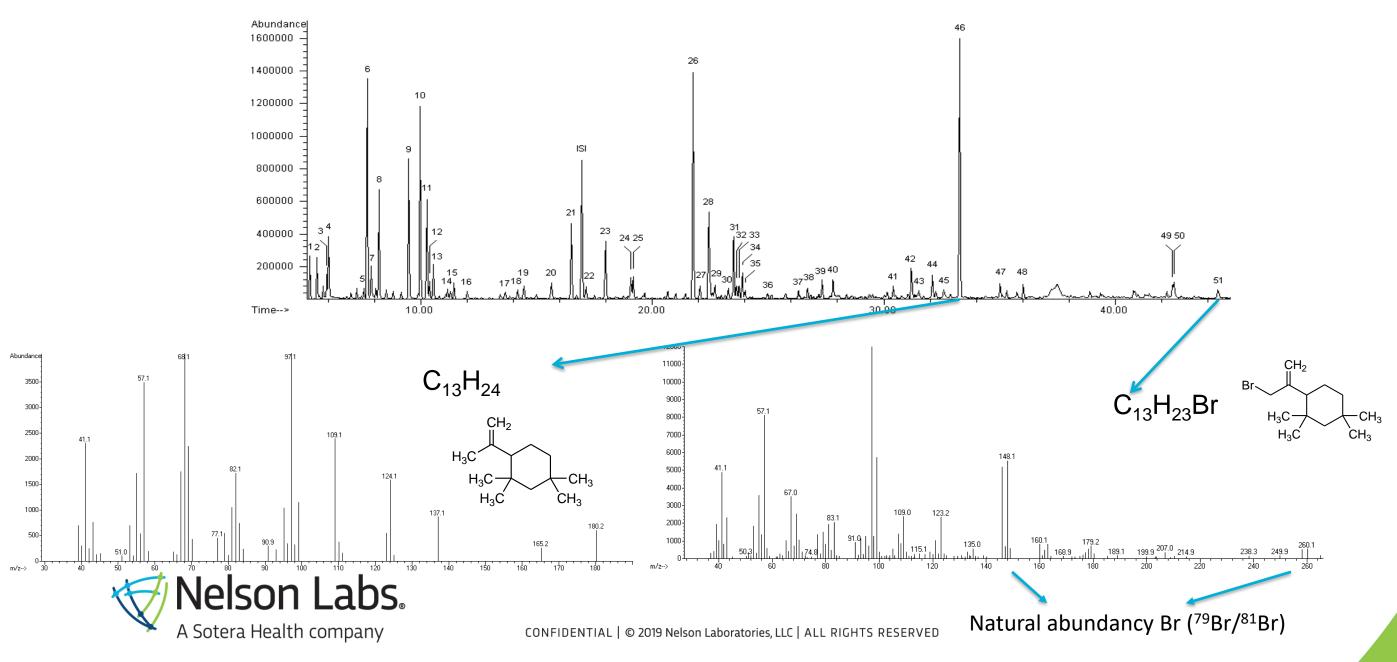
#### 3. Rubber Oligomers: Detection & Identification





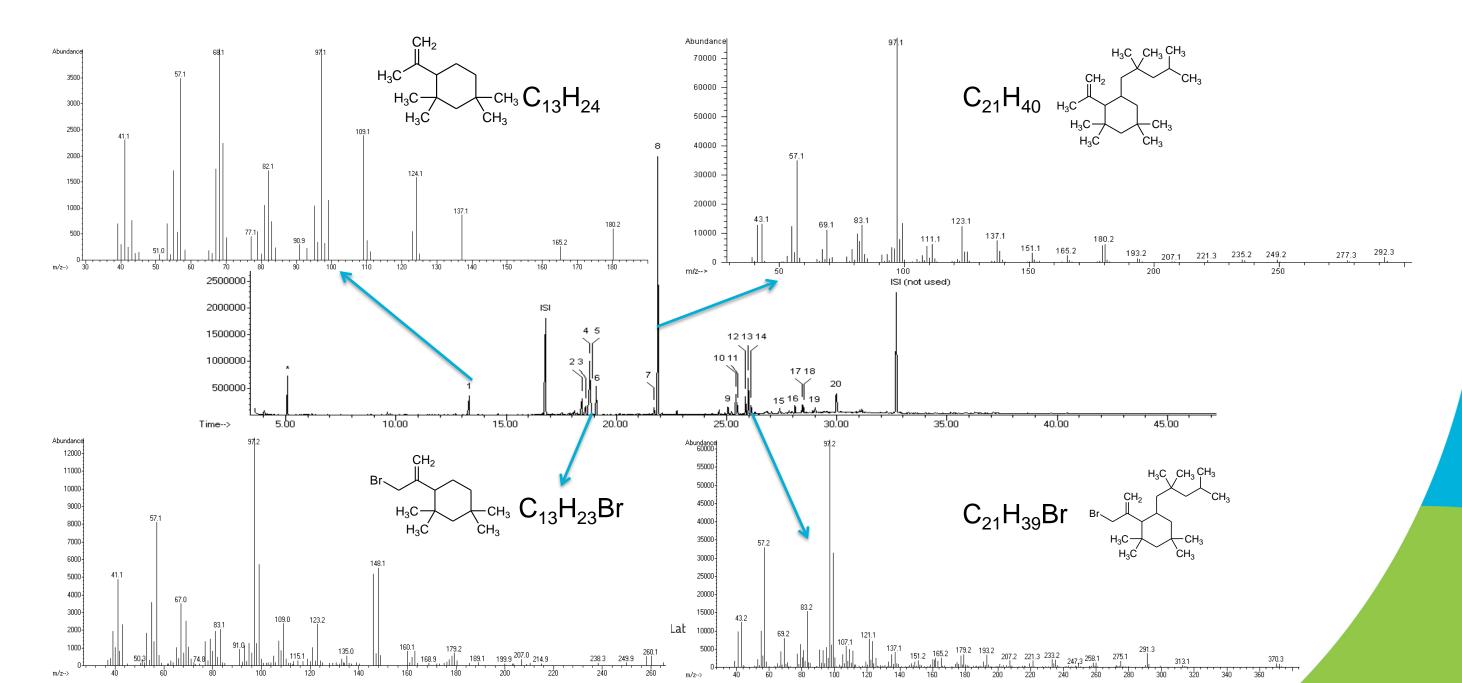
#### **3. Rubber Oligomers – Detection & Identification**

#### Headspace GC/MS on Bromobutyl Rubber Closure:



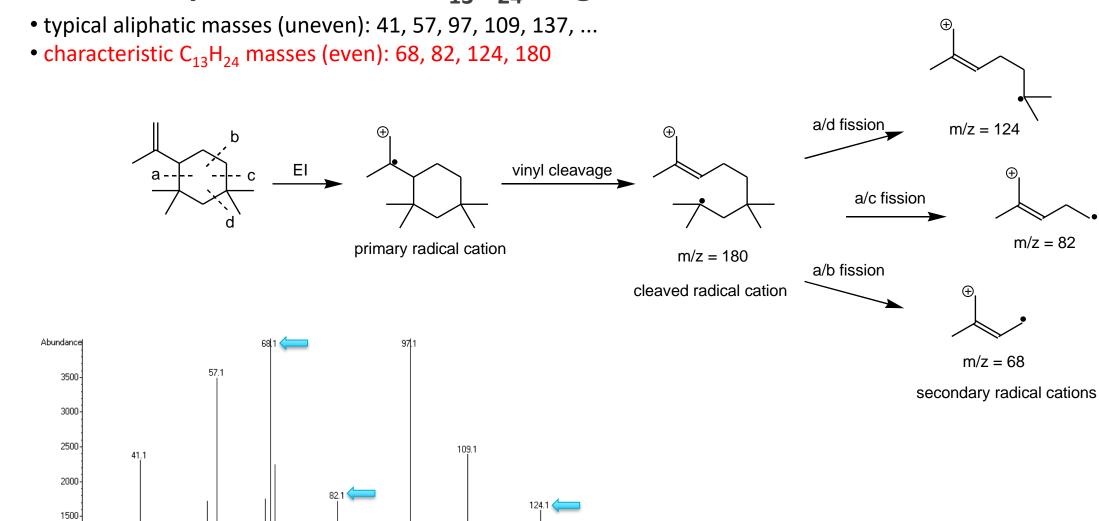
#### **3. Rubber Oligomers – Detection & Identification**

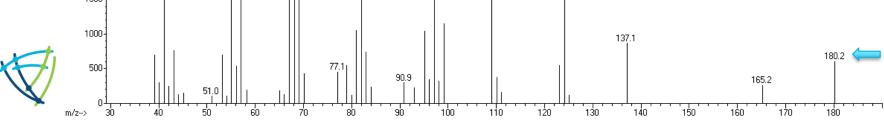
#### **GC/MS** on IPA extract of Bromobutyl Rubber Closure:



#### **3.** Rubber Oligomers – Detection & Identification

#### GC/MS Mass Spectrum of the C<sub>13</sub>H<sub>24</sub> Oligomer





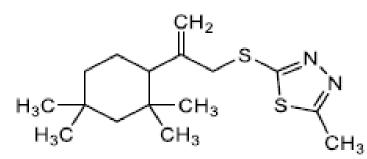


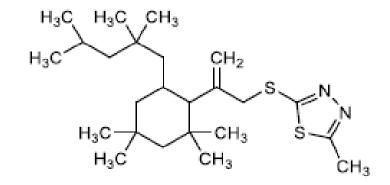


**Reactions with API's** 

#### Allyl Bromide/Chloride: alkylating agent (nucleophilic substitution)

Adduct Formation of an Small Molecule API adducts with the  $C_{13}H_{23}Br$  and  $C_{21}H_{39}Br$  oligomers

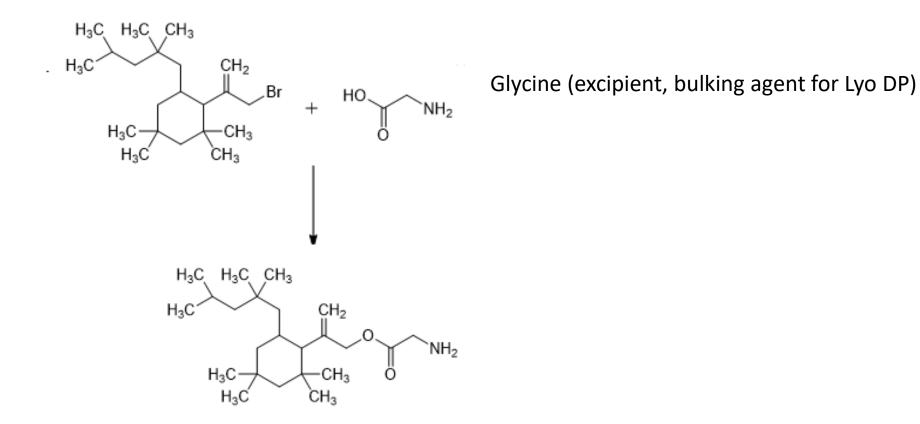






#### **Reactions with Excipients**

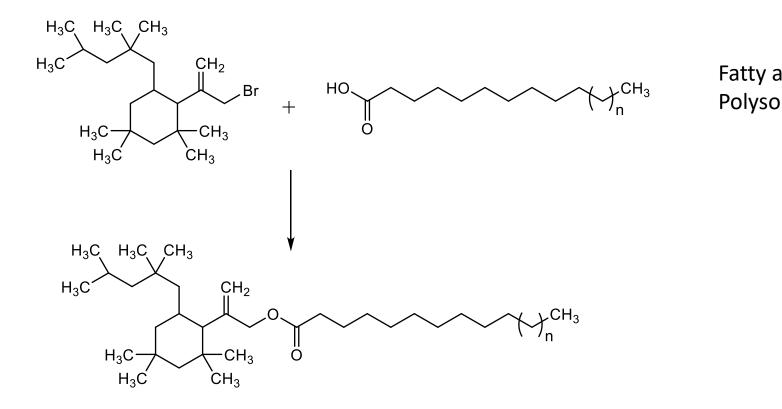
C21H39Br rubber oligomer source leachable from a rubber stopper





#### **Reactions with Excipient Impurities**

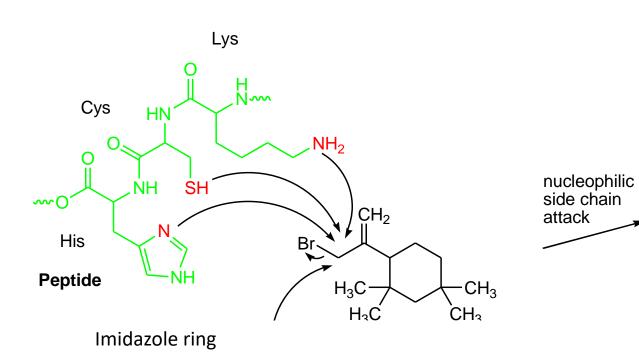
C21H39Br rubber oligomer source leachable from a rubber stopper

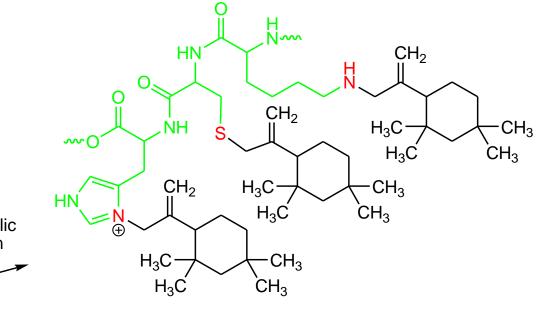




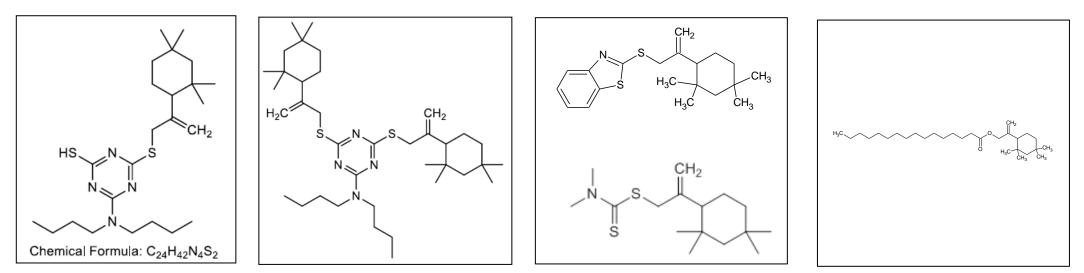
Fatty acids source – Polysorbate 20

#### **Reactions with peptides and proteins**





#### Reactions with other rubber ingredients, formed during the manufacturing process



6-(Dibutylamino)-1,3,5-triazine-2,4-dithiol

N-cyclohexyl-2benzothiazole sulfenamide

Palmitic Acid

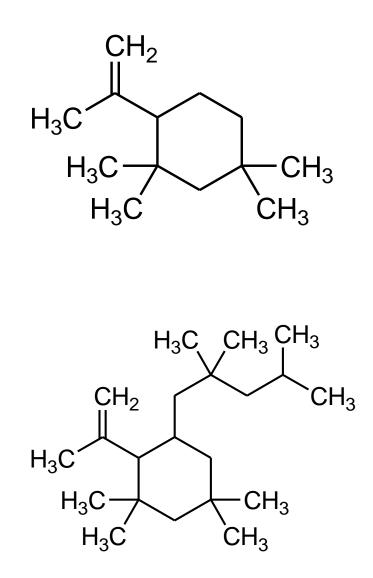
Tetramethylthiuram disulfide



 $C_{13}H_{23}Br / C_{21}H_{39}Br$  With Antioxidants ESBO • • ...









C<sub>13</sub>H<sub>24</sub> and C<sub>21</sub>H<sub>40</sub> Oligomers

- Considered as
  - Cyclic aliphatic hydrocarbon compounds
  - One double bond

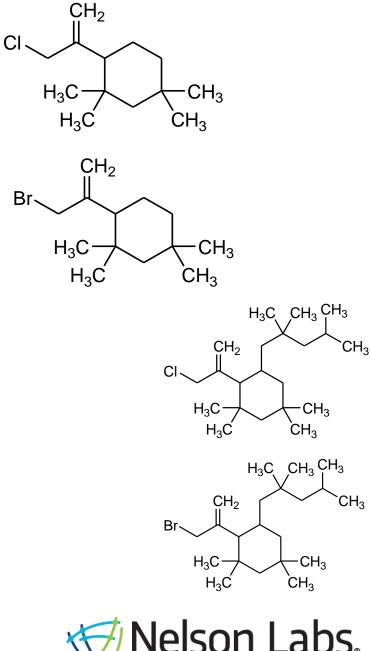
•Structure Activity Relationship (SAR) Assessment:

#### **NO IMMEDIATE CONCERN**

• As no experimental data / Literature data is known about the toxicity of these compounds, a lot of Pharma companies:

- <u>Rely on the result of a SAR assessment</u> to perform a tox evaluation
- <u>Conclude</u> that these compounds are of <u>low Concern</u>

#### erform a tox evaluation <u>Concern</u>





• Considered as

• HALOGENATED Cyclic Aliphatic Hydrobarbon compounds (Allyl Halide)

- Alkylating Agents (reactivity!)
- One double bond

• Structure Activity Relationship (SAR) Assessment: CARCINOGENICITY **SENSITIZATION** 

 As no experimental data / Literature data is known about the toxicity of these compounds, a lot of Pharma companies:

- <u>Rely on the result of a SAR assessment</u> to perform a tox evaluation
- <u>Conclude</u> that these compounds are of <u>High Concern</u>

#### For potential Carcinogenic compounds or Sensitizers:

SCT: 1.5 µg/day (FDA Position for potential Carcinogens) for Chronic Treatments 5 μg/day (FDA Position for potential Sensitizers) for non-Chronic Treatments



NELSON LABS ORIGINAL FINAL REPORT 1 (for Sponsor) **PROJECT NUMBER TE191483** STUDY NUMBER 19-B5589-G1 Nelson Labs Europe Sponsor Contact Piet Christiaens Address Romeinsestraat 12 3001 Leuven Belgium Test Item Arrival Date 06 Feb 2019 Experimental Starting Date 01 Aug 2019 Experimental Completion Date 08 Aug 2019 Study Plan INT/B-Microtox/19/0002 rev 00 Test Facility Nelson Labs NV Address Romeinsestraat 12 3001 Leuven Belgium

SALMONELLA TYPHIMURIUM REVERSE MUTATION ASSAY on Rubber oligomer  $C_{13}H_{23}Br$ 

#### GLP COMPLIANCE STATEMENT

This study meets the technical requirements of the study plan. The study described in this report was conducted in accordance with the OECD Principles of Good Laboratory Practice, except for the characterization of the test item. The study was performed under supervision of the Study Director. The study was conducted according to the procedures herein described and this report represents a true and accurate record of the results obtained.

#### Availability of Rubber Oligomers at Nelson Labs

Allows a correct assessment of the true toxicological risk of these leachables FINAL GLP REPORT: 17-00231-G2

MURINE LOCAL LYMPH NODE ASSAY with Pre-Screen - OECD 429

Test Article

21 CFR Part 58 Compliance Good Laboratory Practice for Nonclinical Laboratory Studies

> Final Report Date 9/11/2018

Study Director Radhika Devalaraja, Ph.D.

#### NELSON AMES REPORTS (Mutagenicity) for CH<sub>13</sub>H<sub>23</sub>Br and C<sub>13</sub>H<sub>23</sub>Cl



#### **NELSON LOCAL LYMPH NODE ASSAY (LLNA)** REPORT for CH<sub>13</sub>H<sub>23</sub>Br

# e-Screen – OECD 429

## Thank you

# Questions? InfoEurope@nelsonlabs.com +32 16 40 04 84





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